

Course Name						
<b>MATERIALS PHYSICS</b>						
<b>Code</b>	<b>Semester</b>	<b>Local Credits</b>	<b>ECTS Credits</b>	<b>Course Implementation, Hours/Week</b>		
				<b>Theoretical</b>	<b>Tutorial</b>	<b>Laboratory</b>
MET 246E	4	2	3	2	-	-
<b>Department/Program</b>		Metallurgical and Materials Engineering				
<b>Course Type</b>		Required		<b>Course Language</b>		ENGLISH
<b>Course Prerequisites</b>		None				
<b>Course Category by Content, %</b>		<b>Basic Sciences</b>	<b>Engineering Science</b>	<b>Engineering Design</b>	<b>General Education</b>	
		30	60	10		
<b>Course Description</b>		Atomic structure and introduction to quantum mechanics; Understanding the Schrödinger Wave Equation, Wave nature of matter; Generation of X ray; Atomic structure and magnetism; Electromagnetic spectrum; Sources of light; Colour and appearance; Reflection, scattering, diffraction; The quantum mechanical description of atomic binding; the crystalline state and amorphous state; Elastic compliance and stiffness constants; Electrical polarization of ionic crystals; Piezo electricity; Modes of vibrations; Phonons; theory of heat capacity, temperature, thermal conductivity, thermal expansion; Free electron and band gap theory; Superconductivity				
<b>Course Objectives</b>		<ol style="list-style-type: none"> <li>1. To make the importance and role of the materials science in the engineering applications understood.</li> <li>2. To teach the quantum physics theories related with the properties and structure of materials and to make the students understand the strong relations between them.</li> <li>3. To give a physics background for understanding the classification of engineering materials with respect to their properties</li> </ol>				
<b>Course Learning Outcomes</b>		After completing this course the student will be able to understand: <ol style="list-style-type: none"> <li>1. Basic concept of quantum mechanics</li> <li>2. The electronic structure of atoms and electronic properties of materials</li> <li>3. Electromagnetic spectrum, sources of light and definition of colour</li> <li>4. The quantum mechanical description of atomic binding, elasticity, lattice vibration and phonons</li> <li>5. Energy and heat concept</li> <li>6. Atomic background of magnetism</li> <li>7. Behaviour of ionic crystals under stress and deformation</li> </ol>				
<b>Textbook</b>		<ul style="list-style-type: none"> <li>• Fredriksson H., Akerlind U., "Physics of Functional Materials," Wiley 2008, ISBN: 978-0-470-51757-4.</li> <li>• Livingston J.D. , "Electronic Properties of Engineering Materials," Wiley 1999, ISBN: 978-0-471-31627-5</li> <li>• Hummel R.E., "Electronic Properties of Materials", 3<sup>rd</sup> Ed., Springer 2005, ISBN No: 0-387-95144-X.</li> <li>• White M.A., "Properties of Materials", Oxford University Press 1999, ISBN No: 978-0195113310.</li> </ul>				
<b>Other References</b>		<ul style="list-style-type: none"> <li>• Kasap S.O., "Principles of Electrical Engineering Materials and Devices", Revised Edition, McGraw – Hill 2000, ISBN No: 0-07-116471-5.</li> <li>• Neamen D.A., "Semiconductor Physics and Devices: Basic Principles", 3<sup>rd</sup> ed., McGraw-Hill 2003, ISBN No: 0-07-119862-8</li> </ul>				
<b>Homework &amp; Projects</b>						
<b>Laboratory Work</b>		none				
<b>Computer Use</b>						
<b>Other Activities</b>						
<b>Assessment Criteria</b>		<b>Activities</b>		<b>Quantity</b>	<b>Effects on Grading, %</b>	
		<b>Midterm Exams</b>		MIN 1	30	
		<b>Quizzes</b>		2	10	
		<b>Homework</b>		-	-	
		<b>Projects</b>		-	-	
		<b>Term Paper/Project</b>		-	-	
		<b>Laboratory Work</b>		-	-	
		<b>Other Activities</b>		-	-	
<b>Final Exam</b>				1	60	

**COURSE PLAN**

Weeks	Topics	Course Outcomes
1	<b>Atomic Structure, Binding and Introduction to Quantum Mechanic:</b> (Early ideas of Atomic structure, Wave-particle duality, Introduction to quantum Mechanic, Understanding the Schrödinger Wave Equation, Wave nature of matter)	1
2	<b>Atomic Structure, Binding and Introduction to Quantum Mechanic:</b> (Early ideas of Atomic structure, Wave-particle duality, Introduction to quantum Mechanic, Understanding the Schrödinger Wave Equation, Wave nature of matter)	1
3	<b>Electron configuration of atoms:</b> (The electronic structure of atoms and periodic table, Electron configuration of transition metals, Quantum mechanics and energy levels, Generation of X ray)	1, 2
4	<b>Electrical Conduction in Solids:</b> (Free electron theory, Band gap theory, Fermi-Dirac Equation, Semiconducting, Superconductivity)	2
5	<b>Electrical Conduction in Solids:</b> (Free electron theory, Band gap theory, Fermi-Dirac Equation, Semiconducting, Superconductivity)	2
6	<b>Optical aspects of matter:</b> (Electromagnetic spectrum, Sources of light, Color and appearance, Refraction and dispersion, Reflection, Scattering, Diffraction, Polarization in optics)	3
7	<b>Optical aspects of matter:</b> (Electromagnetic spectrum, Sources of light, Color and appearance, Refraction and dispersion, Reflection, Scattering, Diffraction, Polarization in optics)	3
8	<b>Crystal and amorphous structures:</b> (The bonding of atoms, The quantum mechanical description of atomic binding, The crystalline state, Amorphous state)	4
9	<b>Lattice vibration and phonons:</b> (Modes of vibrations and Phonons)	4
10	<b>Energy and Heat:</b> (Heat Capacity, theory of heat capacity, temperature, thermal conductivity, quantum and classical theories in heat, thermal expansion)	5
11	<b>Energy and Heat:</b> (Heat Capacity, theory of heat capacity, temperature, thermal conductivity, quantum and classical theories in heat, thermal expansion)	5
12	<b>Magnetism and Electromagnetism:</b> (Atomic background of magnetism, Induction, Electromagnetic waves)	6
13	<b>Ionic Crystals:</b> (Electrical Polarization of ionic crystals, Behaviour of ionic crystals under stress and deformation, Ferroelectric crystals, Piezo electricity)	7
14	<b>Elasticity in Crystals:</b> (Elastic Compliance and Stiffness Constants, Determination of elastic constants, Elastic Waves in Crystals)	4

**Relationship between the Course and METALLURGICAL AND MATERIALS ENGINEERING Curriculum**

	Program Outcomes	Level of Contribution		
		1	2	3
1	Ability to apply the knowledge of mathematics, science and engineering principles to solve problems in metallurgical and materials engineering (ABET:a)			x
2	Ability to characterize materials using standard and/or self designed experimental methods and to evaluate the results (ABET:b)	x		
3	Ability to design a system or a process, taking into consideration of the desired specifications, quality, ethics and environment. (ABET:c)			
4	Ability to communicate both orally and in the written form and to take part in, and provide leadership of the teams in the elucidation of engineering problems; (ABET:d, g)			
5	Ability to define, formulate and solve engineering problems in the development, production, processing, protection and usage of engineering materials. (ABET:e)			x
6	An understanding of professional and ethical responsibilities(ABET:f)			
7	An understanding of current/contemporary issues and impact of engineering solutions in broad cultural, national and global levels;. (ABET:h, j)	x		
8	A comprehension of the nature of engineering progress closely linked with the development of new materials and production processes. An ability to engage in life-long learning and a recognition of its necessity (ABET:i)		x	
9	Ability to use essential tools and techniques of modern engineering in the development, production, processing, protecting and surface treatment of the existing and new engineering materials. (ABET:k)			x

1: Little, 2. Partial, 3. Full

**Course relationships with major elements of the field and material classes**

		Level of Contribution		
		1	2	3
<b>MAJOR ELEMENT OF THE FIELDS</b>	STRUCTURE			x
	PROPERTIES			x
	DESIGN EXPERIMENT/ANALYSE DATA			
	PROCESSING			
	COST/PERFORMANCE			
	QUALITY/ENVIRONMENT			
	DESIGN PROCESS OR PRODUCT	x		
<b>MATERIAL CLASSES</b>	METAL			x
	CERAMICS			x
	POLYMERS		x	
	COMPOSITES	x		

1: Little, 2. Partial, 3. Full

<b>Prepared by</b> Assoc. Prof. Dr. Kürşat KAZMANLI	<b>Date</b> March, 2013	<b>Signature</b>
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