

Course Name						
PRINCIPLES OF CORROSION AND CORROSION PROTECTION						
Code	Semester	Local Credits	ECTS Credits	Course Implementation, Hours/Week		
				Theoretical	Tutorial	Laboratory
MET 437E	7	2	3	2	-	-
Department/Program		Metallurgical and Materials Engineering				
Course Type		Required			Course Language	English
Course Prerequisites		None				
Course Category by Content, %		Basic Sciences	Engineering Science	Engineering Design	General Education	
			30	70		
Course Description		Definition and significance of corrosion. Classification of corrosion. Thermodynamic and kinetic principles of electrochemical corrosion. Passivity. Forms of corrosion: uniform corrosion, pitting, crevice, galvanic corrosion, environmentally induced cracking, corrosion types induced by the flow velocity of the environment, effects of metallurgical structure on corrosion, corrosion related damages by hydrogen, erosion and wear. Corrosive environments: atmosphere, soil, water and aqueous environments, concrete, high temperature environments. Principles of corrosion protection: design, change of metal, change of environment, change of interface. Inorganic, metallic and conversion coatings. Organic coatings and inhibitors. Cathodic and anodic protection. Principles of materials selection for corrosion protection.				
Course Objectives		<p>After completing this course, the student will be able to:</p> <ol style="list-style-type: none"> 1. Know the importance and interdisciplinary character of corrosion. 2. To apply basic corrosion knowledge to engineering problems. 3. Understand material-environment and corrosion type relation. 4. Understand the necessity of learning basic principles of corrosion and corrosion protection for material selection and application. 5. Analyse and bring viable engineering solutions to actual corrosion and corrosion protection problems. 				
Course Learning Outcomes		<ol style="list-style-type: none"> 1. Student will learn the effect of the environment in which materials are designed and produced to function, the importance of theoretical knowledge in devising practical solutions to corrosion problems and of material protection for health, security, engineering and environmental reasons. 2. Student will be able to analyse various corrosion related engineering failure problems, to explain them and suggest engineering solutions. 3. Student will learn to cooperate with other people to attack and solve problem and will also learn how to present his solution. 				
Textbook		D. A. Jones, Principles and Prevention of Corrosion, Macmillan Pub. N.York, 1992. ISBN 0-02-946439-0				
Other References		<ol style="list-style-type: none"> 1. L.L. Shreir, R.A. Jarman and G. Burstein (eds), Corrosion Vol.1,2 and 3 (3th Edition) Butter Worth-Heineman, 1994. 2. K. R. Trethewey and J. Chamberlain, Corrosion for Science and Engineering (2nd edition), Longman Scientific and Technical, Longman Group Technical, Essex, England, 1995 				
Homework & Projects						
Laboratory Work						
Computer Use						
Other Activities						
Assessment Criteria			Quantity	Effects on Grading, %		
Activities			-	-		
Midterm Exams			MIN 1	40		
Quizzes			MIN 2	10		
Homework			-	-		
Projects			-	-		
Term Paper/Project			-	-		
Laboratory Work			-	-		
Other Activities			-	-		
Final Exam			1	50		

COURSE PLAN

Weeks	Topics	Course Outcomes
1	Definition, significance and basic principles of corrosion.	1,3
2	Classification of corrosion	2
3	Thermodynamic principles of electrochemical corrosion.	2
4	Electrochemical kinetics of corrosion	2
5	Passivity and breakdown of passivity.	2
6	Forms of corrosion: Uniform corrosion, localized corrosion; galvanic corrosion and concentration cell corrosion	1,2
7	Forms of corrosion: Pitting, and crevice corrosion, environmentally induced cracking	1,2
8	Forms of corrosion: Effects of metallurgical structure on corrosion. Corrosion related damages by hydrogen, erosion and wear.	1,2
9	Corrosion in selected corrosive environments: atmosphere, soil water and aqueous environments	1,2
10	Microbiologically induced corrosion, concrete corrosion High temperature environments	2,3
11	Principles of corrosion protection: design, change of metal, change of environment, change of interface	2
12	Corrosion protection: inorganic, metallic and conversion coatings	2
13	Corrosion protection: organic coatings and inhibitors, cathodic and anodic protection.	2
14	Principles of materials selection for corrosion protection	2,3

Relationship between the Course and Metallurgical and Materials Engineering Curriculum

	Student Outcomes	Level of Contribution		
		1	2	3
1	an ability to identify, formulate, and solve complex engineering problems by applying principles of engineering science and mathematics			X
2	an ability to apply engineering design to produce solutions that meet specified needs with consideration of public health, safety and welfare as well as global, cultural, social, environmental and economic factors		X	
3	an ability to communicate effectively with a range of audiences			
4	an ability to recognize ethical and professional responsibilities in engineering situations and make informed judgements, which must consider the impact of engineering solutions in global, economic, environmental and societal contexts	X		
5	an ability to function effectively on a team whose members together provide leadership, create a collaborative and inclusive environment, establish goals, plan tasks, and meet objectives			
6	an ability to develop and conduct appropriate experimentation, analyse and interpret data, and use engineering judgement to draw conclusions	X		
7	an ability to acquire and apply new knowledge as needed, using appropriate learning strategies		X	

1: Little, 2. Partial, 3. Full

Course relationships with major elements of the field and material classes

		Level of Contribution		
		1	2	3
MAJOR ELEMENT OF THE FIELDS	STRUCTURE			X
	PROPERTIES			X
	DESIGN EXPERIMENT/ANALYSE DATA			
	PROCESSING	X		
	COST/PERFORMANCE	X		
	QUALITY/ENVIRONMENT	X		
	DESIGN PROCESS OR PRODUCT			X
MATERIAL CLASSES	METAL			X
	CERAMICS AND GLASS			
	POLYMERS			
	COMPOSITES	X		
	BIOMATERIALS			

1: Little, 2. Partial, 3. Full

Prepared by Prof. Dr. Kürşat Kazmanlı Assist. Prof. Dr. Cem Örneke	Date January, 2022	Signature
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