

# ISTANBUL TECHNICAL UNIVERSITY – FACULTY OF CHEMICAL & METALLURGICAL ENGINEERING

## DEPARTMENT OF METALLURGICAL AND MATERIALS ENGINEERING



#### Course Name **MATERIALS CHEMISTRY** Course Implementation, Hours/Week Semester **Local Credits ECTS Credits** Code Theoretical **Tutorial** Laboratory **MET 228E** 2 Department/Program Metallurgical and Materials Engineering Course Type Required **Course Language** English Course None **Prerequisites Engineering** General **Basic Sciences Engineering Science Course Category** Design Education by Content, % 30 50 20 This course provides introductory information to explain the material properties via their place in the periodic table. **Course Description** General approach to the surface properties of solids, the synthesis of inorganic materials through the gas, solution and solid reactions, and fundamental properties of materials such as thermal, optic, magnetic, etc by means of the bonding structure of materials, coordination chemistry and common electrochemical behavior of materials. The scope of lecture is to give: the general explanation of material properties in the periodic table. the idea behind the construction of periodic table. the influence of chemical bonding types and complex structures of materials to material properties **Course Objectives** solubility rules, acids, bases, pH, buffers, ligand exchange, redox reactions the examination of amorphous solids, glass, slags, boron and metal borides, metal carbides and metal nitrides, metal oxides, intermetallics, transition metals by taking the structure of materials into account. the electrical conductivity, optical and thermal properties of materials the synthesis of inorganic materials through the gas, solution and solid reactions, the electrochemical behaviors and properties of metals Students who pass the course will have knowledge on: the fundamental principles that underlie materials chemistry, elements and compounds, chemical formulas and reactions and relation between Periodic table **Course Learning** and location of elements **Outcomes** the influence of chemical bonding types and complex structures of materials to their II. properties III. the structure of materials, the complex compounds IV. inorganic molecules, liquids and solutions ٧. structural solid state chemistry VI. the chemistry of inorganic surfaces VII. synthesis of inorganic material VIII. electrochemical behavior of materials The Inorganic Chemistry Of Materials, Paul J. van der PUT, Prenum Press, NY, ISBN 0-306 **Textbook** 45731-8-2007 Course Hand outs Materials Chemistry, Bradley D. Fahlman, Published by Springer, ISBN 978-1-4020-6119-6 (HB) and ISBN 978-1-4020-6120-2 2008 Other References Harry L Allcock Introduction to Materials Chemistry, 2008, John&Wiley, ISBN 978-0-470-29333-1 **Homework & Projects Laboratory Work Computer Use** Other Activities Activities Quantity Effects on Grading, % Midterm Exams 1 20 20 Quizzes 4 Homework 2 20 **Assessment Criteria Projects** Term Paper/Project **Laboratory Work** Other Activities

**COURSE PLAN** 

Final Exam

40

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Weeks	Topics	Course Outcomes	
1	The Technology of Materials, The role of chemistry in Materials science, Fundemantal Principles that underlie materials chemistry	I - VIII	
2	The Periodic Table, Fundemantal Principles that underlie materials chemistry, Introduction to elements and compunds, chemical formulas and reactions		
3	<b>The Chemical Bond,</b> Electrons in Atomic Shells, Orbitals: Molecular Orbital and Valence Bond Models, The Coordinative Bond in Complexes Bonding in Ionic Compounds,	1, 11, 111	
4	Inorganic Molecules, Liquids and Solutions, solubility rules, acids, bases, pH, buffers Ligand Exchange, Redox Reactions,	I, IV	
5	Structural Solid State Chemistry, Crystal Chemistry, Amorphous Solids, Glass, Slags, Boron and, Metal Borids, Carbides and Nitrides, metal Oxides, Intermetallics, Transition metals,	I, III, IV,V	
6	Structural Solid State Chemistry, Intrinsic Properties, Electron Conductivity, Dielectric Properties, Ion Conductivity, Magnetic, Mechanical, Optical, Chemical, Thermal Properties	I, III, IV,V	
7	Structural Solid State Chemistry, Intrinsic Properties, Electron Conductivity, Dielectric Properties, Ion Conductivity, Magnetic, Mechanical, Optical, Chemical, Thermal Properties	I, III, IV,V	
8	Solid State Reactions, Types of Reactions of Solids, The Chemistry of Oxide, Sulfide and others,	I, III, IV,V	
9	<b>The Chemistry of Inorganic Surfaces</b> , Surface Chemistry, Inorganic Colloids, Converting Solids by Reaction with a Gaseous Recactant, Chemical Vapor Deposition,	IV, VI	
10	<b>The Chemistry of Inorganic Surfaces</b> , The Pyrolylise, Powder Synthesis, High-Temperature Corrosion Surface Modification by Immobilization of Molecules	IV, VI	
11	Synthesis of Inorganic Material, Inorganic Synthesis, Solid State Reactions Synthesis from Liquids, Preparation from Melts,	II, VII	
12	Synthesis of Inorganic Material, Hydrothermal Processes, Sol-Gel Method, A Gas-Phase Techniques, Physical Vapor Deposition, Chemical Vapor Deposition, Plasma Synthesis	II, VII	
13	Electrochemical behavior of materials, material and water interaction, Eh-pH, corrosion, electrochemical reduction and dissolution	I, VIII	
14	Different types of materials and Materials in Advanced Technology		

Relationship between the Course and METALLURGICAL AND MATERIALS ENGINEERING Curriculum

			Level of Contribution	
	Program Outcomes	1	2	3
1	Ability to apply the knowledge of mathematics, science and engineering principles to solve problems in metallurgical and materials engineering (ABET:a)			Х
2	Ability to characterize materials using standard and/or self designed experimental methods and to evaluate the results (ABET:b)			
3	Ability to design a system or a process, taking into consideration of the desired specifications, quality, ethics and environment. (ABET:c)			
4	Ability to communicate both orally and in the written form and to take part in, and provide leadership of the teams in the elucidation of engineering problems; (ABET:d, g)			
5	Ability to define, formulate and solve engineering problems in the development, production, processing, protection and usage of engineering materials. (ABET:e)			Х
6	An understanding of professional and ethical responsibilities(ABET:f)			
7	An understanding of current/contemporary issues and impact of engineering solutions in broad cultural, national and global levels;. (ABET:h, j)			
8	A comprehension of the nature of engineering progress closely linked with the development of new materials and production processes. An ability to engage in life-long learning and a recognition of its necessity (ABET:i)		Х	
9	Ability to use essential tools and techniques of modern engineering in the development, production, processing, protecting of the existing and new engineering materials. (ABET:k)		Х	

## 1: Little, 2. Partial, 3. Full

Course relationships with major elements of the field and material classes

		Level of Contribution		 n
		1	2	3
	STRUCTURE			х
	PROPERTIES			х
MAJOR ELEMENT OF	DESIGN EXPERIMENT/ANALYSE DATA			
THE FIELDS	PROCESSING	Х		
THE FIELDS	COST/PERFORMANCE	Х		
	QUALITY/ENVIRONMENT	Х		
	DESIGN PROCESS OR PRODUCT		Х	
	METAL			х
MATERIAL CLASSES	CERAMICS			х
WATERIAL CLASSES	POLYMERS			
	COMPOSITES			

# 1: Little, 2. Partial, 3. Full

Prepared by	Date	Signature
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