

Course Name							
MASS AND ENE	RGY BALA	NCE Local Credits		Course Implementation, Hours/Week			
Code	Semester	•	ECTS Credits	Theoretical	Tutoria		
MET 248E	4	2	4	2	-	-	
Department/Prog	gram M	etallurgical and Mate	rials Engineering				
Course Type	-	equired	v v	e language	English		
Course Prerequi		ET 215E		3	3		
Course Category		Basic Sciences	Engineering Scienc	e Engineeri	ng Design	General	
by Content, %	/	Dasie Obieniecs		Engineen	ng Design	Education	
			80 %	20	%		
Course descripti	ion Si re er M ch el si	coichiometry; atomic actants, oxidation ar ror, precision, accura aterial Balances; con nemical reaction, r ectrometallurgical a multaneous material	of Units and Conversic and molecular mas ad reduction. Sampling acy and repeatability, n nservation of mass, ma ecycling & by-pass and electrothermic and energy balances,	s, chemical ed and Measurem neasurement of ass balance ana circuits, Ener energy balanc	quations, exc ents Procedu weight, press alyses, systen gy Balances es, staged	ress and limiting res; description o ure, flow rate, etc ns with or withou ; heat balance heat balances	
Course Objectiv	es vi ut vi	th numerical examp ilized in metal produ	etallurgical reactors. e fundamental concepts bles from the existing action Technologies are nes the background of	industrial appli covered within	cations. Almo	ost all processes ork of this course	
Course learning outcomes	1. 2. 3. 4. 5. 6. 7. 8.	Ability to design requirements. Ability to decide, for An extensive educ Aim for students to Aim for students to Ability of studen equipments for en Ability to design a humanity, protecti	owledge of mathemath a system, a product ormulize and solve eng cation for understanding o understand the impor- o be aware of recent ar ts to use necessary gineering applications. and process a system, on of the nature and g the recent necessariti	ct component a nineering probler g engineering so tance of life-time nd modern subje techniques, sh a product and/ for considering	and process ns. plutions global e learning and cts. kills and mo for a process resources in	lly and socially. I learn that ability dern engineering for the benefit o the most efficien	
Textbook		.A. Fine and G.H. Ge	eiger, <u>Handbook on Ma</u>	terial and Energ			
			es, A publication of TM		·· -		
Other references	•	Company. Butts, <u>Metallurgical</u> V. Aytekin, <u>Metalurj</u>	.K. Toner, <u>Conservatio</u> <u>Problems</u> , McGraw-Hil <u>i Problemleri</u> , İTÜ Matb <u>allurgical Engineering</u> ,	l, 1943. aası, 1978.			
Homework & Pre		II homework problem	s are to be handed-in		y are assigne	d. Homework	
Laboratory work	р		d as a source for exam				
Computer use	В	eing able to work wit	h computer programs I	MS Word and M	S Excel		
Other activities	N	ONE					
				Quantity	Effects o	n grading, %	
Assessment crit	eria H P	ctivities lidterm exam luiz omework roject erm Paper/Project		- 1 3 -	1	- 25 % 15 % - -	
		aboratory Work		-		-	
		ther Activities		-		-	
	F	inal exam		1	4	15 %	



	COURSE PLAN				
Weeks	Topics				
1	Dimensions, System of Units and Conversion Factors; molar units, density, concentration.	1			
2	Stoichiometry; atomic and molecular mass, chemical equations	1			
3	Excess and limiting reactants, oxidation and reduction	1			
4	Sampling and Measurements Procedures; description of error, precision, accuracy and repeatability, measurement of weight, pressure, flow rate, etc	1-5			
5	Sampling and Measurements Procedures; description of error, precision, accuracy and repeatability, measurement of weight, pressure, flow rate, etc	1-5			
6	Material Balances; conservation of mass, mass balance analyses	1-3			
7	Material Balances; conservation of mass, mass balance analyses	1-3			
8	Mass balance analyses, systems with or without chemical reaction	1-3			
9	Recycling & by-pass circuits	1-8			
10	Recycling & by-pass circuits	1-8			
11	Energy Balances; heat balance, electrometallurgical and electrothermic energy balances	1-8			
12	Energy Balances; heat balance, electrometallurgical and electrothermic energy balances	1-8			
13	Differential Heat balances, simultaneous material and energy balances, process analysis.	1-8			
14	Examples of materials and energy balances for metallurgical reactors	1-8			

Relationship between the Course and Metallurgical and Materials Engineering Curriculum

	Program Outcomes		Level of Contribution		
		1	2	3	
1	Ability to apply the knowledge of mathematics, science, and engineering principles to solve problems in metallurgical and materials engineering (ABET:a)			Х	
2	Ability to characterize materials using standard and/or self designed experimental methods and to evaluate the results (ABET:b)				
3	Ability to design a system or a process, taking into consideration of the desired specifications, quality, ethics and environment (ABET:c)				
4	Ability to communicate both orally and in the written form and to take part in, and provide leadership of the teams in the elucidation of engineering problems (ABET:d, g)				
5	Ability to define, formulate and solve engineering problems in the development, production, processing, protection and usage of engineering materials (ABET:e)			х	
6	An understanding of professional and ethical responsibilities (ABET:f)				
7	An understanding of current/contemporary issues and impact of engineering solutions in broad cultural, national and global levels (ABET:h, j)				
8	A comprehension of the nature of engineering progress closely linked with the development of new materials and production processes. An ability to engage in life-long learning and a recognition of its necessity (ABET:i)				
9	Ability to use essential tools and techniques of modern engineering in the development, production, processing, protecting of the existing and new engineering materials (ABET:k)		x		

1: Little, 2: Partial, 3: Full

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Course relationships with major elements of the field and material classes

			Level of Contribution	
		1	2	3
	STRUCTURE		Х	
	PROPERTIES			Х
MAJOR ELEMENT OF THE	DESIGN EXPERIMENT/ANALYSE DATA		Х	
FIELDS	PROCESSING	Х		
FIELDS	COST/PERFORMANCE	Х		
	QUALITY/ENVIRONMENT		Х	
	DESIGN PROCESS OR PRODUCT		Х	
	METAL			Х
MATERIAL CLASSES	CERAMICS		Х	
WAIERIAL CLASSES	POLYMERS			
	COMPOSITES			
1: Little, 2: Partial, 3: Full				

 Prepared by
 Date
 Signature

 Prof. Dr. Cüneyt ARSLAN
 March 2013